STANDARD OPERATING PROCEDURE

Dry Ice

Zhou Lab, Institute for Environmental Genomics

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<u>Minimum Personal Protective Equipment Required:</u> Cryogenic gloves, eye protection, skin and body protection (long pants, lab coat, and closed toed shoes)

<u>Risks</u>: Asphyxiation when used or stored in poorly ventilated or tightly enclosed areas; burns or frostbite; explosions if in a closed container.

- Dry ice is the solid form of carbon dioxide and will sublime (vaporize directly to the gas state) at a temperature of -78.5C (-109.3F) or higher.
 - Dry ice will sublimate about 5 to 10 pounds every 24 hours (blocks last longer) in a typical storage cooler.
 - One pound of dry ice produces about 250 liters of gaseous carbon dioxide
- Normal air is composed of 78% nitrogen, 21% oxygen, and only 0.04% carbon dioxide.
 - Concentrations of CO₂ greater than 0.5% (5000 ppm) can become dangerous.
 - Carbon dioxide is denser than air and may accumulate in low, confined spaces with poor ventilation.

Special Handling:

- \checkmark Use or dispense only in well-ventilated areas.
- \checkmark Never transport in the passenger area of a vehicle
- ✓ Never place dry ice inside an ultra-low freezer or other enclosed space!

Protocol/Procedure:

- General use of dry ice
 - Dry ice is used as a cooling agent and is shipped with some reagents or samples received in the lab.
 - Never touch dry ice with bare skin; use tongs or cryogenic gloves to move ice
 - If you want to keep the dry ice for later use, store it in the box it arrived in with the lid loosely on. Do not store the dry ice in a confined area, in freezers or rooms without ventilation
 - Use dry ice in open spaces with good ventilation
 - If at any point a person working with or near dry ice begins to feel unwell, have rapid breathing or cyanosis, they must leave the area immediately
 - To dispose of unneeded dry ice, let the unused portion sublimate in a well-ventilated area.
 - NEVER dispose of dry ice in a sink, the trash, or in any other device or container. The dry ice can destroy the structure because of the temperature difference.
- Shipping dry ice
 - If needed for out-going shipments, dry ice can be purchased in the chemistry stockroom at SLSRC.

- Ask Joy or Lindsay for an account number and bring a Styrofoam cooler to hold the dry ice
- Never transport in the passenger area of a vehicle.
 - If it must be transported via a vehicle, place in trunk or a location apart from the driver
 - Make sure there is good ventilation in the driver's compartment in case of leakage of CO₂ gas into the driver's compartment.
- \circ Dry ice must be packaged in containers that allow the release of CO₂ gas
- If shipping via FedEx or UPS, a dry ice sticker must be placed on the package. These stickers can be obtained from Vanessa in the storeroom.